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forming the precipitate
silver iodide. chemical
equation: $KI(aq) + AgNO_3(aq)$
 $\rightarrow KNO_3(aq) + AgI(s)$ complete

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Net ionic equation: $K^+(aq) + I^-(aq) + 2 Ag^+(aq) + NO_3^-(aq) \rightarrow 0 K^+(aq) + NO_3^-(aq) + AgI(s)$

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sulfur atom ($A_r = 32$) weighs

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twice as much as an oxygen atom ($A_r = 16$), 32 g of sulfur atoms and 16 g of oxygen atoms both contain the same number of atoms.

Answer: 32 g 9.2 (a) 1 mole of H atoms = 1 g 1 mole of H₂ molecules = $2 \times 1 = 2$ g.

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(b) 1 mole of N atoms = 14 g

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names for the following:

Solution a) Since there are
four similar monodentate
ligands. So tetra will be

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Cu and it's oxidation Read
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