

Light Waves Chem Worksheet 5 1 Answer Key

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5-1 **LightWaves.pdf - Light Waves Name Chem Worksheet 5-1** ...

In what part of the electromagnetic spectrum are these waves? Light Waves Name _____ Chem Worksheet 5-1 $m = 1 \times 103 \text{ mm}$ $1 \text{ m} = 1 \times 106 \mu\text{m}$ useful equations $c = \lambda \nu$ $c = 3.00 \times 108 \text{ m/s}$ $1 \text{ m} = 1 \times 10^9 \text{ nm}$ $1 \text{ m} = 1 \times 10^{10} \text{ \AA}$ $c = 3.00 \times 108 \text{ m/s}$ Violet Light ho r tewav l ng and higher frequency Red Light Longer wavelength and lower frequency V I B G Y O R 400 500 600 700 750 nm wavelength (m) 1020 10-9 10-7 10-5 10-3 10-1 101 103 1018 1016 1014 1012 1010 108 106 104 10-11 am rays x-rays ultr ...

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Light Waves Name _____ Chem Worksheet 5-1. $1 \text{ m} = 1 \times 103 \text{ mm}$ $1 \text{ m} = 1 \times 106 \mu\text{m}$. useful equations. $c = \lambda \nu$ $c = 3.00 \times 108 \text{ m/s}$ Shorter wavelength (λ) $1 \text{ m} = 1 \times 109 \text{ nm}$ $1 \text{ m} = 1 \times 10^{10} \text{ \AA}$ $c = 3.00 \times 108 \text{ m/s}$. Violet Light. ν and higher frequency (ν)

1. Draw a wave below and label the following parts: peak ...

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speed of light equation. $c = 3.0 \times 10^8 \text{ m/s}$ and the equation with it is $c = \lambda \nu$ $h = h \nu$ and can be manipulated and isolate different parts to get necessary answer. $\lambda = c / \nu$ energy equation. $E = h \nu$ where h is planks constant in JS of (6.626×10^{-34}) and ν is frequency in s^{-1} and e is in J.

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Radio Waves Worksheets - Kiddy Math

5. What is the frequency of a $7.43 \times 10^{-5} \text{ m}$ wave? 6. What is the frequency of a 2.600 cm wave? 7. What is the wavelength of a $4.34 \times 10^{15} \text{ s}^{-1}$ wave? 8. What is the frequency of a 2.110×10^7 wave? 9. What is the frequency of a $5.6 \times 10^{10} \text{ ?m}$ wave? 10. What is the wavelength of 109.6 MHz wave? 11. What is the energy of a $7.66 \times 10^{14} \text{ Hz}$ wave? 12.

Chemistry Worksheet – Wavelength, frequency, & energy of ...

The diagram below shows white light dispersed into its different colours by a prism: Light travels at $300\,000 \text{ km/s}$. It takes light 8.5 minutes to reach the Earth from the Sun. Apart from the electromagnetic waves, there are also matter waves, like water waves and sound waves. These travel at a much lower speed.

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Planck's Equation Name Chem Worksheet 5-2. © John Erickson, 2005 $W = 2hc/\lambda$ Planck's Eq. $E = h \nu$ $h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$ useful equations $c = \lambda \nu$ $c = 3.00 \times 108 \text{ m/s}$ $1 \text{ m} = 1 \times 109 \text{ nm}$ $1 \text{ kJ} = 1000 \text{ J}$. example. Light with a wavelength of 525 nm is green. Calculate the energy in joules for a green light photon. - find the frequency: $c = \lambda \nu$

Planck's Equation Name Chem Worksheet 5-2

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